# Structure of the Product Formed via Regiospecific Cyclopropanation of exo-6-Methoxycarbonyltricyclo[5.2.1.0 ${ }^{2,6}$ ]deca-4,8-dien-3-one with Dimethylsulfoxonium Methylide (Corey's Reagent) <br> By William H. Watson* and Ante Nagl $\dagger$ <br> Department of Chemistry, Texas Christian University, Fort Worth, Texas 76129, USA <br> and Alan P. Marchand* and G. Madhusudhan Reddy <br> Department of Chemistry, University of North Texas, PO Box 5068, Denton, Texas 76203, USA 

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#### Abstract

Compound (2a), methyl 3-oxotetracyclo[6.2.1.0 ${ }^{2,7} .0^{4,6}$ ]undec-9-ene-7-carboxylate, $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{3}$, $M_{r}=218.25$, triclinic, $\quad P \overline{1}, \quad a=6.470(1), \quad b=$ 7.523 (1),$\quad c=12.967$ (1) $\AA, \quad \alpha=74.50$ (1),$\quad \beta=$ 88.97 (1), $\gamma=64.50(1)^{\circ}, \quad V=545.5$ (1) $\AA^{3}, \quad Z=2$, $D_{x}=1.329 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda($ Mo $K \alpha)=0.71073 \AA, \quad \mu=$ $0.88 \mathrm{~cm}^{-1}, F(000)=232, T=300, R=0.0448$ for 1587 independent reflections. Compound $2(a)$ is composed of three fused five-membered rings and a three-membered ring. Molecular mechanics calculations indicate the system has about $268 \mathrm{~kJ} \mathrm{~mol}^{-1}$ of strain energy. Although substitution of the norbornene moiety is asymmetric, the bond lengths across the ring system are statistically equivalent. The cyclopropane bond lengths range from 1.479 (2) to 1.512 (3) $\AA$. The double-bond system of the norbornene moiety is folded along the $\mathrm{C}(9)-\mathrm{C}(10)$ bond and deviates from planarity by $9(2)^{\circ}$. The H atoms are bent away from the methylene bridge.


Experimental. Under the reaction conditions shown in the scheme, a single product was obtained in $80 \%$ yield;

however, the reaction product might possess either structure (2a) or (2b). A single crystal of compound

[^0](2a), m.p. $354-355 \mathrm{~K}$, was obtained by fractional recrystallization from pentane. A colorless crystal of dimensions $0.23 \times 0.40 \times 0.43 \mathrm{~mm}$ was mounted on a Nicolet $R 3 M / \mu$ update of a $P 2_{1}$ diffractometer; data collected in the $\omega$ scan mode ( $3.0 \leq 2 \theta \leq 50^{\circ}$ ), variable scan rate ( $4-29.3^{\circ} \mathrm{min}^{-1}$ ), graphite-monochromated Mo $K \alpha$ radiation; lattice parameters from a leastsquares refinement of 25 reflections ( $25 \cdot 3 \leq 2 \theta \leq$ $29.7^{\circ}$ ), intensity statistics were consistent with space group $P \overline{1}$; monitored reflections $\overline{3} \overline{1} 2$ and $\overline{1} 4 \overline{2}$ show no significant changes in intensities; 1920 independent reflections measured ( $-7 \leq h \leq 7,-8 \leq k \leq 8,0 \leq$ $l \leq 15), \quad 1587 \geq 3 \sigma(I)$; Lorentz-polarization corrections, $\psi$-scan-based empirical absorption correction (transmission factors $0.816-0.858$ ); structure solved by direct-methods techniques, block-cascade leastsquares refinement, H atoms refined with isotropic thermal parameters; $R=0.0448, w R=0.0569$ for 201


Fig. 1. Drawing of compound (2a) with thermal ellipsoids drawn at the $35 \%$ probability level. H atoms are represented by spheres of arbitrary size.
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Table 1. Atomic coordinates $\left(\times 10^{4}\right)$ and equivalent isotropic thermal parameters $\left(\AA^{2} \times 10^{3}\right)$
$U_{\text {eq }}$ defined as one third of the trace of the orthogonalized $U_{l j}$ tensor.

| $x$ |  |  |  |  |
| :--- | ---: | ---: | ---: | ---: |
| $r$ | $y$ | $U_{\mathrm{cu}}$ |  |  |
| $\mathrm{C}(1)$ | $-228(3)$ | $3327(3)$ | $6527(1)$ | $48(1)$ |
| $\mathrm{C}(2)$ | $403(3)$ | $1440(2)$ | $7555(1)$ | $40(1)$ |
| $\mathrm{C}(3)$ | $2533(3)$ | $-496(2)$ | $7537(1)$ | $42(1)$ |
| $\mathrm{C}(4)$ | $3017(3)$ | $-2313(3)$ | $8468(1)$ | $54(1)$ |
| $\mathrm{C}(5)$ | $1795(3)$ | $-2091(3)$ | $7425(1)$ | $47(1)$ |
| $\mathrm{C}(6)$ | $-724(3)$ | $-1192(3)$ | $7366(2)$ | $53(1)$ |
| $\mathrm{O}(6)$ | $-1896(3)$ | $-1977(2)$ | $7174(2)$ | $88(1)$ |
| $\mathrm{C}(7)$ | $-1670(3)$ | $944(3)$ | $7508(1)$ | $47(1)$ |
| $\mathrm{C}(8)$ | $-3245(3)$ | $2641(3)$ | $6486(2)$ | $60(1)$ |
| $\mathrm{C}(9)$ | $-1978(3)$ | $2177(3)$ | $5529(2)$ | $64(1)$ |
| $\mathrm{C}(10)$ | $-190(3)$ | $2552(3)$ | $5559(1)$ | $57(1)$ |
| $\mathrm{C}(11)$ | $-2826(3)$ | $4452(3)$ | $6558(2)$ | $59(1)$ |
| $\mathrm{C}(12)$ | $625(3)$ | $2107(3)$ | $8531(1)$ | $48(1)$ |
| $\mathrm{O}(12)$ | $-804(3)$ | $2568(3)$ | $9137(1)$ | $81(1)$ |
| $\mathrm{O}(13)$ | $2619(2)$ | $2185(2)$ | $8628(1)$ | $60(1)$ |
| $\mathrm{C}(13)$ | $3045(4)$ | $2821(4)$ | $9528(2)$ | $76(1)$ |

parameters and 1587 reflections ( $R=0.0544$ and $w R=0.0586$ for all 1920 reflections), $S=1.528$, $(\Delta / \sigma)_{\text {max }}=0.035$; largest peaks in the final difference map of 0.19 and $-0.24 \mathrm{e}^{-3} ; \sum w\left(\left|F_{o}\right|-\left|F_{c}\right|\right)^{2}$ minimized with $w=\left[\sigma^{2}\left(F_{o}\right)+0.00074 F_{o}{ }^{2}\right]^{-1}$. All computer programs supplied by Nicolet Instrument Corporation (1986) for Desktop 30 Microeclipse and Nova 4/C configurations; atomic scattering factors and anomalous-dispersion corrections from International Tables for X-ray Crystallography (1974). The MM2 molecular mechanics program was used to calculate strain energies (Allinger \& Yuh, 1980). Fig. 1 is a drawing of compound ( $2 a$ ). Table 1 lists atomic positional parameters, and Table 2 gives interatomic distances and valence angles.*

Related literature. endo-Tricyclo[5.2.1.0 $0^{2,6}$ dec-8-en-3ones are of interest as intermediates in the synthesis of polyquinane natural products (Smith, Marchand, Suri \& Jin, 1986; Marchand \& Vidyasagar, 1988). Corey \& Chaykovsky (1965) reported the use of dimethylsulfoxonium methylide in cyclopropanation reactions. The results of a study of [4+2] Diels-Alder cycloaddition of substituted homobenzoquinones to cyclopentadiene have been reported recently (Adam, Kliem, Lucchini, Peters, Peters \& von Schnering, 1988). The crystal structures of the two related Diels-Alder cycloadducts (3) and (4) were reported. The deviations

[^1]Table 2. Bond lengths $(\AA)$ and valence angles $\left({ }^{\circ}\right)$

| $\mathrm{C}(1)-\mathrm{C}(2) \quad 1$ | 1.571 (2) | $\mathrm{C}(1)-\mathrm{C}(10) \quad 1$ | 1.514 (3) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(1)-\mathrm{C}(11) \quad 1$. | 1.528 (2) | $\mathrm{C}(2)-\mathrm{C}(3) \quad 1$ | 1.520 (2) |
| $\mathrm{C}(2)-\mathrm{C}(7) \quad 1$. | 1.545 (3) | $\mathrm{C}(2)-\mathrm{C}(12) \quad 1$ | 1.512 (3) |
| $\mathrm{C}(3)-\mathrm{C}(4) \quad 1$ | 1.479 (2) | $\mathrm{C}(3)-\mathrm{C}(5) \quad 1$ | 1.509 (3) |
| $\mathrm{C}(4)-\mathrm{C}(5) \quad 1$ | 1.512 (3) | $\mathrm{C}(5)-\mathrm{C}(6) \quad 1$ | 1.466 (2) |
| C.5)-O(6) $\quad 1$ | 1.207 (3) | $\mathrm{C}(6)-\mathrm{C}(7) \quad 1$ | 1.516 (3) |
| $\mathrm{C}\left({ }^{\prime}\right)-\mathrm{C}(8) \quad 1$ | 1.565 (2) | $\mathrm{C}(8)-\mathrm{C}(9) \quad 1$ | 1.510 (3) |
| $\mathrm{C}(8)-\mathrm{C}(11) \quad 1$. | 1.525 (4) | $\mathrm{C}(9)-\mathrm{C}(10) \quad 1$. | 1.307 (4) |
| $\mathrm{C}(12)-\mathrm{O}(12) \quad 1$. | 1.198 (2) | $\mathrm{C}(12)-\mathrm{O}(13) \quad 1$ | 1.326 (3) |
| $\mathrm{O}(13)-\mathrm{C}(13) \quad 1$. | 1.446 (3) |  |  |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(10)$ | 106.9 (1) | $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(11)$ | 99.9 (1) |
| $\mathrm{C}(10)-\mathrm{C}(1)-\mathrm{C}(11)$ | $100 \cdot 1$ (2) | $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | 115.4 (1) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(7)$ | $102 \cdot 2$ (1) | $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(7)$ | $106 \cdot 3$ (2) |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(12)$ | 108.3 (1) | $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(12)$ | 110.9 (1) |
| $\mathrm{C}(7)-\mathrm{C}(2)-\mathrm{C}(12)$ | 113.6 (1) | $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | 115.1 (1) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(5)$ | 108.8 (1) | $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(5)$ | $60 \cdot 8$ (1) |
| $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | $60 \cdot 6$ (1) | $\mathrm{C}(3)-\mathrm{C}(5)-\mathrm{C}(4)$ | 58.6 (1) |
| $\mathrm{C}(3)-\mathrm{C}(5)-\mathrm{C}(6)$ | 108.0 (1) | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 115.4 (2) |
| $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{O}(6)$ | 125.4 (2) | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(7)$ | 110.0 (2) |
| $\mathrm{O}(6)-\mathrm{C}(6)-\mathrm{C}(7)$ | 124.5 (2) | $\mathrm{C}(2)-\mathrm{C}(7)-\mathrm{C}(6)$ | 106.6 (1) |
| $\mathrm{C}(2)-\mathrm{C}(7)-\mathrm{C}(8)$ | 102.9 (1) | $\mathrm{C}(6)-\mathrm{C}(7)-\mathrm{C}(8)$ | 112.2 (2) |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(9)$ | 106.5 (1) | $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(11)$ | 100.4 (2) |
| $\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{C}(11)$ | 99.9 (2) | $\mathrm{C}(8)-\mathrm{C}(9)-\mathrm{C}(10)$ | $107 \cdot 8$ (2) |
| $\mathrm{C}(1)-\mathrm{C}(10)-\mathrm{C}(9)$ | 107.6 (2) | $\mathrm{C}(1)-\mathrm{C}(11)-\mathrm{C}(8)$ | $93 \cdot 8$ (1) |
| $\mathrm{C}(2)-\mathrm{C}(12)-\mathrm{O}(12)$ | 125.9 (2) | $\mathrm{C}(2)-\mathrm{C}(12)-\mathrm{O}(13)$ | 111.1 (1) |
| $\mathrm{O}(12)-\mathrm{C}(12)-\mathrm{O}(13)$ | ) 123.0 (2) | $\mathrm{C}(12)-\mathrm{O}(13)-\mathrm{C}(13)$ | ) 116.8 (2) |

from planarity of the double-bond systems in norbornenes and sesquinorbornenes have been reported previously (Watson, 1983).

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[^1]:    * Lists of H -atom coordinates, anisotropic thermal parameters and structure factors have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 51610 ( 16 pp .). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

